An Unusual Base-catalysed Rearrangement of an α -Ethoxycarbonylethyl (or α -Cyanoethyl) Cyclic β -Keto Ester

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Treatment of the title compounds (3), (7), and (8) with potassium t-butoxide afforded the unusual rearrangement products (5), (9), and (10).

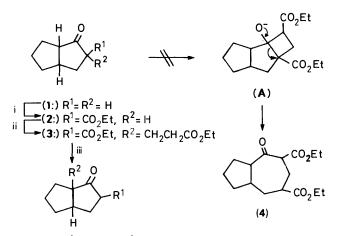
It is generally accepted that, under basic conditions, a compound with an active methylene at the γ -position to the carbonyl function, *e.g.* (3), may undergo a zip reaction to afford two ring-expanded products *via* the four-membered intermediate (A) (see Scheme 1), instead of retro-Michael cleavage.¹ In continuation of studies on ring-expansion reactions,² we have found a novel, unexpected rearrangement.

Ethoxycarbonylation of bicyclo[3.3.0]octan-2-one (1)³ with ethyl cyanoformate afforded (2) (70%) which upon Michael addition with ethyl acrylate gave a single product (3). Treatment of the latter with Bu'OK in Me₂SO at 25 °C gave three components two of which were identified as the retro-Michael cleavage product (2) (20%) and unchanged (3) (12%). The ¹H n.m.r. spectrum of the third component (22%), obtained as an inseparable stereoisomeric mixture, showed remarkable differences from that of (3). A newly formed quaternary carbon was observed at 58.92 p.p.m. [(3) 61.80 p.p.m.] in its ¹³C offresonance n.m.r. spectrum, which suggests that this is not the expected compound (4), but the rearrangement product (5). The g.c.m.s. analysis also offered additional proof (fragment peak: 225 ($M^+ - CO_2Et$), 123 [225 - CH₂CH₂CO₂Et] for the assigned structure (5). Decarboxylation⁴ of (5) with NaCl $Me_2SO-H_2O/150$ °C afforded (6) (66%), the structure of which was supported by spectroscopic data g.c.-m.s., 270 MHz ¹H n.m.r., ¹³C n.m.r., and long range 2D ¹H-¹³C COSY spectrum).†

This unprecedented rearrangement was also observed in the cyclopentanone systems (7) and (8) which on treatment with Bu^tOK in Me₂SO afforded (9) (57%) and (10) (47%) respectively. These were converted into the corresponding decarboxylated compounds (11) and (12) \ddagger (Scheme 2). With the sixand seven-membered ring systems (13) and (14), the rearrangement failed to take place, only retro-Michael cleavage

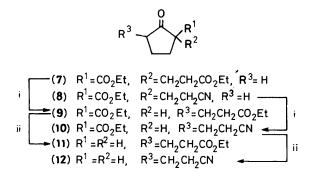
[†] All compounds gave satisfactory spectroscopic data. For selected data: (6); $δ_{\rm H}({\rm CDCl}_3)$ 1.24 (t, J7.1 Hz, 3-H), 1.45—2.50 (m, 15 H), 4.11 (q, J7.1 Hz, 2-H); $δ_{\rm C}({\rm CDCl}_3)$ 14.2 (q), 24.9 (t), 25.3 (t), 30.3 (t), 31.0 (t), 33.4 (t), 35.9 (t), 37.5 (t), 46.3 (d), 59.3 (s), 60.3 (t), 173.4 (s), and 224.5 (s); $v_{\rm max}$.(neat) 2 950 and 1 730 cm⁻¹; m/z (e.i-m.s.) 224 (M^+), 196, 179, and 123 (Found: M^+ , 224.1407. Calc. for C₁₃H₂₀O₃: 224.2992). ‡ Chemical and spectroscopic characteristics are in full agreement with

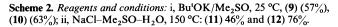
[‡] Chemical and spectroscopic characteristics are in full agreement with that of the authentic samples prepared by alkylation of pyrrolidin-1ylcyclopentene with ethyl acrylate and acrylonitrile, respectively, see G. Stork, A. Brizzolara, H. Landesman, J. Szmuszhovicz, and R. Terrell, J. Am. Chem. Soc., 1963, 85, 207.



iv (5:) R¹ = CO₂Et, R² = CH₂CH₂CO₂Et (6:) R¹ = H, R² = CH₂CH₂CO₂Et

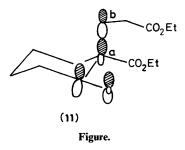
Scheme 1. Reagents and conditions: i, EtOCOCN-lithium di-isopropylamine (LDA)/THF, -78 °C, 70%; ii, Bu'OK/Bu'OH, ethyl acrylate, 25 °C, 65%; iii, Bu'OK/Me₂SO, 25 °C, 22%; iv, NaCl-Me₂SO-H₂O, 150 °C, 66%.





products (15) and (16) being isolated (Scheme 3). This finding suggests that the reaction may involve an intramolecular rearrangement. We believe that the quasi-axial bond of the side chain in the five-membered ring (7) plays an important role in suppressing the retro-Michael reaction.* In addition, this bond can adopt an appropriate trajectory⁵ for a reverse [3.3] sigmatropic rearrangement⁶ to the proximal carbonyl to

* It seems to be reasonable to assume that methoxycarbonylethyl (or cyanoethyl) function in six- and seven-membered rings occupies the axial configuration, and the overlap between the carbonyl π -orbital and the σ *-orbital of C_a-C_b should then facilitate cleavage of the C_a-C_b bond (see Figure).

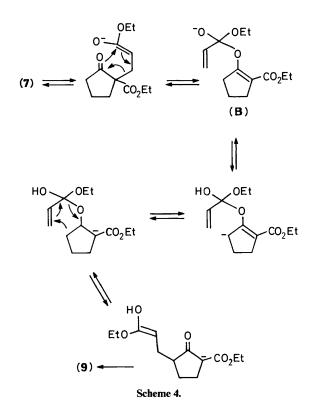






(13) n = 6, $R^1 = CO_2Et$, $R^2 = CH_2CH_2CO_2Me(CN)$ (15) n = 6, $R^1 = CO_2Et$, $R^2 = H$ (14) n = 7, $R^1 = CO_2Me$, $R^2 = CH_2CH_2CO_2Me(CN)$ (16) n = 7, $R^1 = CO_2Me$, $R^2 = H$

Scheme 3.



generate the intermediate (**B**). Thus, subsequent rearrangement 7 as shown in Scheme 4 may afford (9).

Experimental

1B-Ethoxycarbonylethyl-3-ethoxycarbonyl-5BH-bicyclo-[3.3.0] octan-2-one (5).—Compound (3) (180 mg, 0.61 mmol) in Me_2SO (1.5 ml) was added dropwise with stirring to Bu^tOK (82 mg, 0.73 mmol) in dry Me₂SO (2 ml) at 20 °C. Stirring was continued for 40 min at 20 °C after which the reaction was quenched by acetic acid (0.25 ml). The reaction mixture was diluted with brine and extracted with AcOEt. The AcOEt extract was washed with 5% aqueous NaHCO3 and brine, dried, and evaporated under reduced pressure to leave an oil, which was purified by flash column chromatography on silica gel. The fraction eluted with 5% AcOEt in hexane (v/v) afforded (2) (24 mg, 20%), (5) (40 mg, 22%) as a colourless oil, and recovered (3) (22 mg, 12%). For (5): v_{max} (neat) 3 400, 1 730, and 1 680 cm⁻¹; δ(CDCl₃) 4.21, 4.20 (3:2) (2 H, q, J 7.1 Hz), 4.11 (2 H, q, J 7.1 Hz), 3.37 (3 H, t, J 8.6 Hz), 2.80-1.61 (13 H, m), 1.30, 1.26 (3:2) (3 H, t, J 7.1 Hz), and 1.24 (3 H, t, J 7.1 Hz); 296 (M⁺), 250, 196, 179, and 122.

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